Element cards for Part 1

Hydrogen

Atomic mass: 1.008
Description: odorless gas, very flammable
Compounds: H₂O, HCl

Li Lithium

Atomic mass: 6.941

Description: soft gray metal, reacts

with water

Compounds: Li₂O, LiCl

Be

Beryllium

Atomic mass: 9.012 Atomic

Description: gray metal Compounds: BeO, BeCl₂

B

Boron

Atomic mass: 10.81

Description: gray metalloid,

semiconductor

Compounds: B2O3, BCl3

 \mathbf{C}

Carbon

Atomic mass: 12.01

Description: black solid (graphite) or transparent crystal (diamond)

Compounds: CO2, CCl4

N

Nitrogen

Atomic mass: 14.01

Description: odorless gas, rather

unreactive

Compounds: NH3, NCl3

O

Oxygen

Atomic mass: 16.00

Description: odorless gas, very

reactive

Compounds: H2O

F

Fluorine

Atomic mass: 19.00

Description: yellowish gas,

extremely reactive

Compounds: HF, NaF, CaF2

Na Sodium

Atomic mass: 22.99

Description: soft gray metal, reacts

vigorously with water Compounds: Na₂O, NaCl

Mg Magnesium

Atomic mass: 24.31

Description: gray metal, flammable

Compounds: MgCl2, MgO

Al Aluminum

Atomic mass: 26.98 Description: silvery metal Compounds: AlCl₃, Al₂O₃

Si

Silicon

Atomic mass: 28.09

Description: gray metalloid,

semiconductor

Compounds: SiCl4, SiO2

P

Phosphorus

Atomic mass: 30.97

Description: white, red, or black solid, spontaneously flammable

Compounds: PH3, PCl3, PCl5

S

Sulfur

Atomic mass: 32.07

Description: yellow solid powder

Compounds: H2S, SCl2

Cl

Chlorine

Atomic mass: 35.45 Description: greenish gas, extremely reactive

Compounds: HCl, NaCl, CaCl2

K

Potassium

Atomic mass: 39.10

Description: soft gray metal, reacts violently with water Compounds: K₂O, KCl

Ca

Calcium

Atomic mass: 40.08

Description: hard silvery metal,

flammable

Compounds: CaCl2, CaO

As

Arsenic

Atomic mass: 74.92

Description: gray metalloid

Compounds: AsH3, AsCl3, AsCl5

Se

Selenium

Atomic mass: 78.96

Description: gray or red solid

Compounds: H2Se, SeCl2

Br

Bromine

Atomic mass: 79.90

Description: red-orange liquid, very

reactive

Compounds: HBr, NaBr, CaBr2

Rb

Rubidium

Atomic mass: 85.47

Description: soft gray metal, reacts

violently with water

Compounds: Rb2O, RbCl

Sr

Strontium

Atomic mass: 87.62

Description: soft silvery metal

Compounds: SrCl2, SrO

In

Indium

Atomic mass: 114.8

Description: soft silvery metal

Compounds: InCl3, In2O3

Sn

Tin

Atomic mass: 118.7

Description: silvery-white metal

Compounds: SnO₂, SnCl₄

Sb	
Antimony	
Atomic mass: 121.8	

Description: bluish-white metalloid,

semiconductor

Compounds: SbH3, SbCl3, SbCl5

Te

Tellurium

Atomic mass: 127.6

Description: silvery-white metalloid, semiconductor Compounds: H₂Te, TeCl₂

Iodine

Atomic mass: 126.9

Description: dark-purple solid,

reactive

Compounds: HI, NaI, CaI2

Element cards for Part 2

Germanium

Atomic mass: 72.59

Description: gray metalloid,

semiconductor

Compounds: GeO2, GeCl4

Gallium

Atomic mass: 69.72

Description: silvery metal, melts at

just above room temperature Compounds: GaCl3, Ga2O3

Element cards for Part 3

He	Ne
Helium	Neon
Atomic mass: 4.003 Description: odorless gas, very unreactive Compounds: none known	Atomic mass: 20.18 Description: odorless gas, very unreactive Compounds: none known
Ar Argon	Kr Krypton
Atomic mass: 39.95 Description: odorless gas, very unreactive Compounds: none known	Atomic mass: 83.80 Description: odorless gas, very unreactive Compounds: none known
Xe Xenon	
Atomic mass: 131.3 Description: odorless gas, very unreactive Compounds: XeF ₆ , XeF ₄	