Directions: Draw **Lewis structures** for the following and label any polar bonds with **polarity arrows/partial charges**.

1. N2F4 (will contain N-N bond instead of a central atom)
2. ClBr
3. N2Cl2 (will contain N-N bond instead of a central atom and Cl is chlorine)
4. AsI3
5. CH2O
6. C2F6 (will contain C-C bond instead of a central atom)
7. SCl2
8. CH3I
9. GeBr3Cl
10. BrO21+